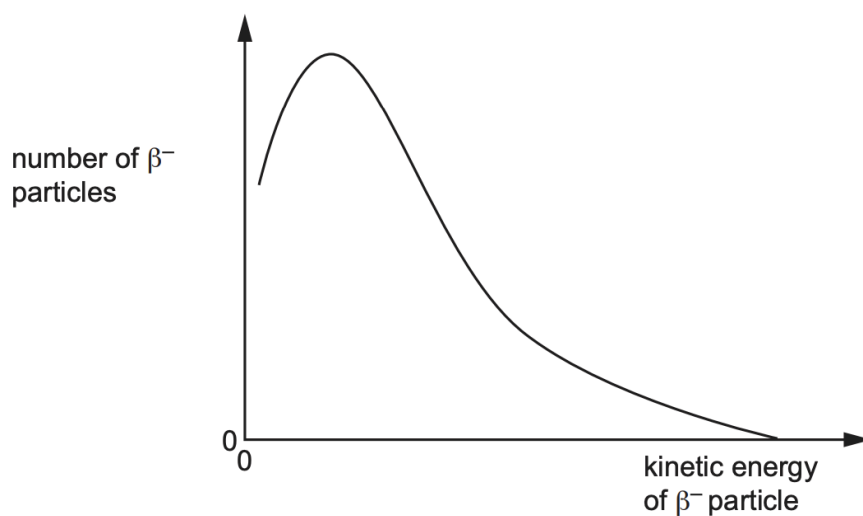


1. An isolated stationary nucleus X decays by emitting an α -particle to form a nucleus Y. Nucleus Y and nucleus Z are isotopes of the same element. Use the principle of conservation of momentum to explain why nucleus Y cannot be stationary immediately after the decay of nucleus X.
 - total momentum before decay is zero OR X has zero / no momentum
 - total momentum after decay must be zero so Y must have equal (and opposite) momentum to α -particle (so cannot be stationary / must have speed/velocity)

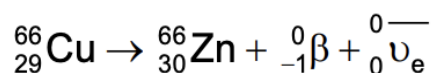
2.

The energy spectrum of the β^- radiation emitted by a sample of this isotope is shown in Fig. 7.1.



Use Fig. 7.1 to explain why other particles apart from the β^- particles must be emitted during this decay.

- the energy of the decay is fixed / constant
 - the energies of the beta particles have a continuous range of values / varies / not constant
 - another particle (antineutrino) must possess the extra / remaining energy
3. State the name of the other particle emitted during β^- decay.
 - Electron antineutrino
 4. Copper isotope ${}^{66}\text{Cu}$ decays to an isotope of Zn. Give the radioactive decay equation for this decay. Include the nucleon and proton numbers of all the particles involved.



NOTE: when asked to quantitatively state the change in charge, say 0 if no change, +1 if increase by 1, -1 if decrease by one, etc.

5. A particle P is composed of an up quark and a down antiquark. Particle P belongs to two classes (groups) of particles. State the names of these two classes.
 - hadron(s)
 - meson(s)

6. Compare an α -particle with a β^+ particle in terms of their masses and charges.
 - mass of α -particle is much greater than β^+ particle
 - both particles are positively charged
 - magnitude of charge on α -particle is twice the charge on β^+ particle

7. State what is meant by a fundamental particle
 - particle with no internal structure / particle which cannot be broken down into anything smaller